Exam Seat No:	Enrollment No:
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C.U.SHAH UNIVERSITY

Wadhwan City

Subject Code :: 5SC01CHC1
Subject Name : Inorganic Chemistry

Summer Examination-2014

Date: 9/06/2014

Time:02:00 To 5:00

Branch/Semester:- M.Sc(Chemistry/II) Examination: Regular

Instructions:-

- (1) Attempt all Questions of both sections in same answer book / Supplementary
- (2) Use of Programmable calculator & any other electronic instrument is prohibited.
- (3) Instructions written on main answer Book are strictly to be obeyed.
- (4) Draw neat diagrams & figures (If necessary) at right places
- (5) Assume suitable & Perfect data if needed

SECTION-I

Q-1	Do as Directed.	(07)
a)	Explain Ferromagnetism	(02)
b)	Write the bowng angle value of water	(02)
c)	Define diamagnetic	(01)
d)	Define electron density.	(01)
e)	Give definition of stereo chemical	(01)
Q-2	Answer the following	(14)
a)	Discuss the pi-bond and delocalized energy of 1,3 butadiene.	(05)
b)	Calculate the magnetic moment values for Fe (III), Cu(II) and Ti(II)	(05)
c)	Find out the ground state term symbol of i) Cu(II), ii) Ni(II), iii) Fe(II)	(04)
0.3	OR TRANS YOUR DE	(1.4)
Q-2	Answer the following	(14)
a)	Give an account on the properties of paramagnetic bodies	(05)
b)	Discuss magnetic susceptibility by Gouy's Method	(05)
c)	Define the following i) Magnetic induction, ii) Van Vleck formula	(04)
Q-3	Answer the following	(14)
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a)	Discuss the stereo chemical applications of the first transition series	(07)
b)	Explain the magnetic application of Lanthanides and Actinides series	(07)
0.1	OR	(1.4)
Q-3	Answer the following in detail.	(14)
a)	Discuss the Huckel π -electron theory and its applications	(08)
b)	Define the following i) Pascal's Constant, ii) Russell-Saunders coupling iii) Anti and ferrimagnetism	(06)

SECTION-II

Q-4	Do as Directed.	(07)
a)	Define chemical shift	(02)
b)	What is Doppler effect?	(02)
c)	What is the full form DART?	(01)
d)	Give equation for recoil energy	(01)
e)	Define quadrupole splitting?	(01)
Q-5	Answer the following	(14)
a)	Write a brief note on application of Mossbauer spectroscopy	(05)
b)	Explain the principle of Mossbauer effect	(05)
c)	How is the EDTA useful reagent in inorganic analysis?	(04)
	OR	
Q-5	Answer the following	(14)
a)	Discuss the Ammonium Vanadate (NH4VO3), Ceric Suphate [Ce(SO4)2] as the reagent	(05)
b)	What is Mossbauer spectroscopy? Draw the diagram of Mossbauer spectrophotometer	(05)
c)	Write a note on reagents i) DMG, ii) Dithiooxamide	(04)
Q-6	Answer the following in detail.	(14)
a)	Write a note on i) Benzidine, ii) Salicylaldoxime, iii) o-phenanthroline inorganic reagent	(06)
b)	Discuss the Potassium Bromate (KBrO3), Potassium Iodate (KIO3) as the useful inorganic reagents	(08)
	OR	
Q-6	Answer the following in detail.	(14)
a)	Write a note on i) Dithiozone, ii) Aluminon iii) Oxine inorganic reagents	(06)
b)	How the inorganic reagents useful for organic reactions and write a note on Anthranilic and rubenic acid.	(08)

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